Simultaneously Predicting Optimum pH Value and Optimum Temperature in Catalytic Reaction of Beta-glucosidase ^{*} 贝塔-葡萄糖苷酶催化反应的最适 pH 值和最适温度的 同时预测

YAN Shao-min¹, SHI De-qiang¹, NONG Hao¹, WU Guang^{2**} 严少敏¹, 师德强¹, 农 浩¹, 吴 光²

(1. State Key Laboratory of Non-food Biomass Enzyme Technology, National Engineering Research Center for Non-food Biorefinery, Guangxi Key Laboratory of Biorefinery, Guangxi Academy of Sciences, Nanning, Guangxi, 530007, China; 2. DreamSciTech Consulting, Shenzhen, Guangdong, 518054, China)

(1. 广西科学院非粮生物质国家重点实验室, 国家非粮生物质能源工程技术研究中心, 广西生物 炼制重点实验室, 广西南宁 530007; 2. 深圳市追梦科技咨询有限公司, 广东深圳 518054)

Abstract: The features of amino acids were used to simultaneously predict optimum pH value and optimum temperature of beta-glucosidases. Firstly, the beta-glucosidases were quantified by different features of amino acids as inputs, and their optimum pH value and optimum temperature were served as outputs; secondly, the training was conducted by means of 20-2 feedforw ard backpropagation neural network; finally, the validation was performed by three approaches, subset validation, jackknife validation, and cross-validation. The results showed that among 24 features of amino acids only 4 features worked in the prediction model and the amino-acid distribution probability as predictor gave better results. Thus, the method developed in this study paved the way towards the prediction of functional parameters of enzymes based on their amino -acid properties.

Key words: amino acid, beta-glucosidase, optimum pH value, optimum temperature, prediction working condition

摘要:利用氨基酸特征同时预测贝塔-葡萄糖苷酶反应的最适 pH 值和最适温度。首先,用不同的氨基酸特征对 贝塔-葡萄糖苷酶进行量化作为输入,以最适 pH 值和最适温度作为输出;然后用 20-2 前馈反向传播的神经网络 进行训练;最后用子集验证、刀切验证和交叉验证三种方法进行验证。结果表明,在 24 个氨基酸特征中只有 4 个特征可以用于模型预测,其中氨基酸分布概率的预测结果优于其它指标,为基于氨基酸属性预测酶的功能参 数提供了方法。

关键词: 氨基酸 贝塔-葡萄糖苷酶 最适 pH 值 最适温度 预测 工作条件 中图法分类号: Q814.9, Q939.9 文献标识码: A 文章编号: 1005-9164(2011) 03-0253-08

In order to speed up enzymatic reaction, it is

作者简介: 严少敏(1958-), 女, 博士, 研究员, 主要从事计算变异学和模 型研究。

* This study was partly supported by Guangxi Academy of Sciences (08 YJ 65 W06) and Guangxi Science Foundation (0907016, 0991013, 0991006Z, 1004606, 1103111, 2010GXNSFF013003 and

2010GXNSFA013046).

* *通讯作者。

广西科学 2011 年 8 月 第 18 卷第 3 期

important to let the enzymatic reaction be at the optimal conditions. Generally the optimal working conditions are so elegant that we have to conduct many expensive and time-consuming experiments to find them. Although these costly optimal working conditions can serve as references to novel enzymes, the inter-relationship among optimal working conditions is oftentimes difficult to manage.

With the fast development in computational bi-

收稿日期: 2011-01-07

ology and bioinformatics, it would be possible to develop computational methods to predict the optimal working conditions for novel enzymes based on previously obtained optimal conditions.

Actually, many studies so far have been directed to the function-structure relationship of proteins including enzymes. However, this relationship has yet to connect with the optimal conditions for enzymatic reactions, and this relationship is more or less related to secondary, tertiary, and quaternary structures, whose determinations are also costly and time-consuming.

Thus, the challenge arises whether we can use very simple features, such as the features of amino acids, to predict optimal conditions of enzymes, because the prediction would help us economically optimize enzymatic reactions.

As β -glucosidase (EC 3. 2. 1.21) can cut the β bond linkage in glucose molecules^[1], it plays an important role in biological processes. More notably, it can degrade celluloses, which gives a great perspective in the fermentation of biomass into biofuels and leads to more efforts to not only search for β -glucosidases but also mutate β -glucosidases^[2]. Consequently, we can find more and more β -glucosidases with annotations of their primary structures but without optimal conditions for enzymatic reactions.

Actually, there are many features of amino acids available^[3,4], which could serve as predictors to predict the optimal conditions for β -glucosidase reactions. However, it is yet to know which feature is really useful. In this study, we attempt to find which of 24 amino-acid features can simultaneously predict optimum pH value and optimum temperature of β glucosidases.

1 Materials and methods

1.1 Data

The data of β -glucosidases (EC 3.2.1.21) were obtained from the Comprehensive Enzyme Information System BRENDA up to October 2010^[5]. Under the functional parameters of pH optimum and temperature optimum, 34 β -glucosidases had their sequence information, and one had documented its mu $\tan t^{[6]}$. Frequently, an enzyme can have several different optimum values of pH and temperature, which was the case for four β -glucosidases^[7]. In total, we found 39 β -glucosidases with their sequence, optimum pH value and optimum temperature in this databank.

1.2 Predictive model

As there may be various linear and nonlinear relationships between the feature of amino acid and optimum pH value and optimum temperature of β glucosidase, we used a 20-2 feedforward backpropagation neural network displayed in Fig. 1 to account for these relationships^[8]. In this model, the first layer contained 20 neurons corresponding to 20 inputs. which could be any piece of features related to 20 types of amino acids in β -glucosidase. The second layer contained two neurons corresponding to two outputs that are optimum pH and optimum temperature. The transfer functions were tan-sigmoid and linear for two layers, and log-sigmoid for output. The training algorithm was the resilient backpropagation, which was the fastest algorithm on pattern recognition in MatLab^[9].

1.3 Features of amino acids

The very basic feature of primary structure of β -glucosidase was its amino-acid composition, and then we had the molecular weights of amino acids (row 2, Table 1). Moreover, we had the features of amino acids related to the spatial properties listed in rows 3 ~ 5 in Table 1^[10, 11], hydrophobic properties listed in rows 6 ~ 10 in Table 1^[12], electronic properties listed in rows 11 ~ 17 in Table 1^[13], and the secondary structure predictions listed in rows 18 ~ 24 in Table 1^[14]. We weighed the values in Table 1 with amino-acid composition because β -glucosidases were different one from another in terms of amino-acid compositions.

Another relatively recent feature about the primary structure was the amino - acid distribution probability, which also represented spatial characteristics on protein^[15~17] and was computed according to the occupancy of subpopulations and partitions^[18], which gives each type of amino acid a distribution probability in an enzyme (T able 2).

Fable 1	Features of amino acids used	as predictors for	predicting optimum	pH value and o	ptimum temperature

Table 1 Fratures of annuo actos used as predictors for predicting optimum pri varue and optimum temperature										
Amino acid	А	R	Ν	D	С	Е	Q	G	Н	Ι
Mass(Dalton)	71.09	156.19	114.11	115.09	103.15	129.12	128.14	57.05	137.14	113.16
Surface area $(Å^2)$	115	225	160	150	135	190	180	75	195	175
Surrace area (II)	110		100	100	100	170	100	, 0	1,0	170
Residue volume(Å ³)	166. 7	88.6	173.4	114.1	111.1	108.5	138.4	143.8	60. 1	153.2
van der Waals volume (Å ³)	67	148	96	91	86	114	109	48	1 18	124
Residue non-polar sur- face area(Å ²)	86	89	42	45	48	69	66	47	129	155
Residue burial (kcal/ mol)	2.15	2.23	1.05	1.13	1.2	1.73	1.65	1.18	2.45	3.88
Side chain burial(kcal∕ mol)	1	1.1	-0.1	-0.1	0	0.5	0.5	0	1.3	2.7
Hydropathy index	4.50	4.20	-0.80	-0.90	- 3 . 50	- 0 . 70	- 1.60	1.80	-3.90	- 3 . 50
Ranking of amino acid polarities	9	15	16	19	7	18	17	11	10	1
pK _a	9.69	9.04	8.8	9.6	10.28	9.67	9.13	9.6	9.17	9.68
$\sigma_{\rm I}$	0.05	-0.26	-0.14	0.51	-0.01	0.68	-0.1	0	-0.01	0.06
$H_M \Delta P H$	0.05	-0.75	-0.2	1.8	-0.01	1.25	-0.07	0	0.21	0.08
$\sigma_{\rm R}$	0	-0.49	-0.06	1.29	0.01	0.57	0.03	0	0.22	0.02
σα	-0.01	-0.08	-0.04	-0.03	-0.03	-0.04	-0.05	0	-0.06	-0.04
б. П	0.05	0.27	-0.56	- 1 77	0.06	-1.14	-0.35	0	-0.58	0.04
٥ _F	0.05	0.27	0.50	0.51	0.00	0.69	0.55	0	0.50	0.04
A_{I}	0.05	0.20	0.24	0.51	0.01	0.08	0.1	0	0.01	0.06
P(a)	142	98	101	67	70	151	111	57	100	108
P(b)	83	93	54	89	119	37	110	75	87	160
P(turn)	66	95	146	156	119	74	98	156	95	47
f(i)	0.06	0.07	0.147	0.161	0.149	0.056	0.074	0.102	0.14	0.043
$f(i \pm 1)$	0.076	0.106	0.11	0.083	0.05	0.06	0.098	0.085	0.047	0.034
f(i+2)	0.035	0 099	0 179	0 191	0 117	0.077	0.037	0.19	0.093	0.013
f(i+2)	0.058	0.095	0.081	0.001	0.128	0.064	0.008	0.152	0.054	0.015
Amino acid	1	V. 005	M	F	D. 120	<u>0.004</u>	<u>0.070</u> Т	W	V	V
A mino aciu	L	ĸ	MI	r	07.12	97.09	101 11	186 12	163 18	00 1/
Mass(Dalton)	113 16	128 17	131 10	1/17 18	G / I /	A / 11A		1001.17.		JJ. 1 4
Mass(Dalton) Surface area $(Å^2)$	113.16 170	128.17 200	131. 19 185	147.18 210	97.12 145	87.08 115	140	255	230	155
Mass(Dalton) Surface area(Å ²) Basidue volume(Å ³)	113.16 170 166.7	128.17 200	131.19 185 162.9	147.18 210 189.9	97.12 145 112.7	87.08 115 89	140	255	230	155 140
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³)	113. 16 170 166. 7	128. 17 200 168. 6	131. 19 185 162. 9	147. 18 210 189. 9	97. 12 145 112. 7	87.08 115 89	140 116. 1	255 227. 8	230 193. 6	155 140
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue per palar surp	113. 16 170 166. 7 124	128. 17 200 168. 6 135	131. 19 185 162. 9 124	147. 18 210 189. 9 135	97.12 145 112.7 90	87.08 115 89 73	101. 11 140 116. 1 93	255 227. 8 163	103. 18 230 193. 6 141	155 140 105
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/	113. 16 170 166. 7 124 122 3. 05	128. 17 200 168. 6 135 164 4, 1	131. 19 185 162. 9 124 137 3. 43	147. 18 210 189. 9 135 194 3. 46	97. 12 145 112. 7 90 124 3. 1	87.08 115 89 73 56 1.4	140 116. 1 93 90 2. 25	255 227. 8 163 236 4. 11	103. 18 230 193. 6 141 154 2. 81	155 140 105 135 3, 38
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol)	113. 16 170 166. 7 124 122 3. 05	128. 17 200 168. 6 135 164 4. 1	131. 19 185 162. 9 124 137 3. 43	147. 18 210 189. 9 135 194 3. 46	97.12 145 112.7 90 124 3.1	 87.08 115 89 73 56 1.4 	140 116. 1 93 90 2. 25	255 227. 8 163 236 4. 11	230 193. 6 141 154 2. 81	155 140 105 135 3. 38
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial(kcal/ mol)	113. 16 170 166. 7 124 122 3. 05 1. 9	128. 17 200 168. 6 135 164 4. 1 2. 9	 131. 19 185 162. 9 124 137 3. 43 2. 3 	147. 18 210 189. 9 135 194 3. 46 2. 3	97.12 145 112.7 90 124 3.1 1.9	 87.08 115 89 73 56 1.4 0.2 	140 116. 1 93 90 2. 25 1. 1	255 227. 8 163 236 4. 11 2. 9	230 193. 6 141 154 2.81 1.6	155 140 105 135 3. 38 2. 2
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial(kcal/ mol) Hydropathy index	113. 16 170 166. 7 124 122 3. 05 1. 9 - 1. 30	128. 17 200 168. 6 135 164 4. 1 2. 9 2. 50	131. 19 185 162. 9 124 137 3. 43 2. 3 - 0. 40	147. 18 210 189. 9 135 194 3. 46 2. 3 - 3. 20	97. 12 145 112. 7 90 124 3. 1 1. 9 - 3. 50	 87.08 115 89 73 56 1.4 0.2 2.80 	140 116.1 93 90 2.25 1.1 1.90	255 227. 8 163 236 4. 11 2. 9 4. 50	230 193. 6 141 154 2.81 1.6 3.80	155 140 105 135 3. 38 2. 2 - 3. 50
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial(kcal/ mol) Hydropathy index Ranking of amino acid polarities	113. 16 170 166. 7 124 122 3. 05 1. 9 - 1. 30 3	128. 17 200 168. 6 135 164 4. 1 2. 9 2. 50 20	131. 19 185 162. 9 124 137 3. 43 2. 3 - 0. 40 5	147. 18 210 189. 9 135 194 3. 46 2. 3 - 3. 20 2	$97.12 \\ 145 \\ 112.7 \\ 90 \\ 124 \\ 3.1 \\ 1.9 \\ -3.50 \\ 13 \\ 13$	 87.08 115 89 73 56 1.4 0.2 2.80 14 	140 116.1 93 90 2.25 1.1 1.90 12	255 227. 8 163 236 4. 11 2. 9 4. 50 6	230 193. 6 141 154 2.81 1.6 3.80 8	155 140 105 135 3. 38 2. 2 - 3. 50 4
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial(kcal/ mol) Hydropathy index Ranking of amino acid polarities pK _a	$ \begin{array}{r} 113.16\\170\\166.7\\124\\122\\3.05\\1.9\\-1.30\\3\\9.6\end{array} $	128. 17 200 168. 6 135 164 4. 1 2. 9 2. 50 20 8. 95	131. 19 185 162. 9 124 137 3. 43 2. 3 - 0. 40 5 9. 21	147. 18 210 189. 9 135 194 3. 46 2. 3 - 3. 20 2 9. 13	$97.12 \\ 145 \\ 112.7 \\ 90 \\ 124 \\ 3.1 \\ 1.9 \\ -3.50 \\ 13 \\ 10.6 \\ $	 87.08 115 89 73 56 1.4 0.2 2.80 14 9.15 	140 116.1 93 90 2.25 1.1 1.90 12 9.1	255 227. 8 163 236 4. 11 2. 9 4. 50 6 9. 39	230 193. 6 141 154 2.81 1.6 3.80 8 9.11	155 140 105 135 3. 38 2. 2 - 3. 50 4 9. 62
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial(kcal/ mol) Hydropathy index Ranking of amino acid polarities pK _a σ_1	$ \begin{array}{r} 113. 16 \\ 170 \\ 166. 7 \\ 124 \\ 122 \\ 3. 05 \\ 1. 9 \\ - 1. 30 \\ 3 \\ 9. 6 \\ 0. 02 \\ \end{array} $	128. 17 200 168. 6 135 164 4. 1 2. 9 2. 50 20 8. 95 - 0. 16	$ \begin{array}{c} 131. 19\\ 185\\ 162. 9\\ 124\\ 137\\ 3. 43\\ 2. 3\\ - 0. 40\\ 5\\ 9. 21\\ 0. 08\\ \end{array} $	147. 18 210 189. 9 135 194 3. 46 2. 3 - 3. 20 2 9. 13 0. 04	$97.12 \\ 145 \\ 112.7 \\ 90 \\ 124 \\ 3.1 \\ 1.9 \\ -3.50 \\ 13 \\ 10.6 \\ 0$	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ \end{array} $	$ \begin{array}{c} 140\\ 140\\ 116.1\\ 93\\ 90\\ 2.25\\ 1.1\\ 1.90\\ 12\\ 9.1\\ -0.05\\ \end{array} $	255 227. 8 163 236 4. 11 2. 9 4. 50 6 9. 39 0. 06	230 193. 6 141 154 2.81 1.6 3.80 8 9.11 0.05	155 140 105 135 3. 38 2. 2 - 3. 50 4 9. 62 0. 01
$\begin{array}{l} M \operatorname{ass}(\operatorname{Dalton}) \\ \operatorname{Surface} \operatorname{area}(\operatorname{\AA}^2) \\ \operatorname{Residue} \operatorname{volume}(\operatorname{\AA}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\operatorname{\AA}^3) \\ \operatorname{rander} W \operatorname{aals} \operatorname{volume}(\operatorname{\AA}^3) \\ \operatorname{Residue} \operatorname{non-polar} \operatorname{surface} \operatorname{area}(\operatorname{\AA}^2) \\ \operatorname{Residue} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Hyd} \operatorname{ropathy} \operatorname{index} \\ \operatorname{Ranking} \operatorname{of} \operatorname{amino} \operatorname{acid} \\ \operatorname{polarities} \\ \operatorname{pK}_a \\ \sigma_1 \\ \operatorname{H}_M \Delta \operatorname{PH} \end{array}$	$ \begin{array}{r} 113.16\\170\\166.7\\124\\122\\3.05\\1.9\\-1.30\\3\\9.6\\0.02\\0.07\\\end{array} $	128. 17 200 168. 6 135 164 4. 1 2. 9 2. 50 20 8. 95 - 0. 16 - 1. 11	$ \begin{array}{r} 131. 19\\ 185\\ 162. 9\\ 124\\ 137\\ 3. 43\\ 2. 3\\ - 0. 40\\ 5\\ 9. 21\\ 0. 08\\ - 0. 04\\ \end{array} $	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06 \end{array} $	$97.12 \\ 145 \\ 112.7 \\ 90 \\ 124 \\ 3.1 \\ 1.9 \\ -3.50 \\ 13 \\ 10.6 \\ 0 \\ 0.1 \\ 0.1$	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ -0.05 \\ \end{array} $	$ \begin{array}{c} 140\\ 140\\ 116.1\\ 93\\ 90\\ 2.25\\ 1.1\\ 1.90\\ 12\\ 9.1\\ -0.05\\ -0.03\\ \end{array} $	255 227. 8 163 236 4. 11 2. 9 4. 50 6 9. 39 0. 06 0. 15	230 193. 6 141 154 2.81 1.6 3.80 8 9.11 0.05 0.02	$ \begin{array}{r} 155 \\ 140 \\ 105 \\ 135 \\ 3. 38 \\ 2. 2 \\ - 3. 50 \\ 4 \\ 9. 62 \\ 0. 01 \\ 0. 09 \\ \end{array} $
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial(kcal/ mol) Hydropathy index Ranking of amino acid polarities pK _a σ_1 H _M Δ PH σ_R	$ \begin{array}{c} 113. 16\\ 170\\ 166. 7\\ 124\\ 122\\ 3. 05\\ 1. 9\\ - 1. 30\\ 3\\ 9. 6\\ 0. 02\\ 0. 07\\ 0. 05\\ \end{array} $	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$	$ \begin{array}{r} 131. 19\\ 185\\ 162. 9\\ 124\\ 137\\ 3. 43\\ 2. 3\\ - 0. 40\\ 5\\ 9. 21\\ 0. 08\\ - 0. 04\\ - 0. 12\\ \end{array} $	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02 \end{array} $	97.12 145 112.7 90 124 3.1 1.9 -3.50 13 10.6 0 0.1 0.1	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ -0.05 \\ -0.02 \\ \end{array} $	$ \begin{array}{c} 140\\ 140\\ 116.1\\ 93\\ 90\\ 2.25\\ 1.1\\ 1.90\\ 12\\ 9.1\\ -0.05\\ -0.03\\ 0.02\\ \end{array} $	255 227. 8 163 236 4. 11 2. 9 4. 50 6 9. 39 0. 06 0. 15 0. 09	$ \begin{array}{c} 103.13\\ 230\\ 193.6\\ 141\\ 154\\ 2.81\\ 1.6\\ 3.80\\ 8\\ 9.11\\ 0.05\\ 0.02\\ -0.03\\ \end{array} $	$ \begin{array}{r} 155 \\ 140 \\ 105 \\ 135 \\ 3. 38 \\ 2. 2 \\ - 3. 50 \\ 4 \\ 9. 62 \\ 0. 01 \\ 0. 09 \\ 0. 08 \\ \end{array} $
$\begin{array}{l} M \operatorname{ass}(\operatorname{Dalton}) \\ \operatorname{Surface} \operatorname{area}(\mathring{A}^2) \\ \operatorname{Residue} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{residue} \operatorname{non-polar} \operatorname{surface} \operatorname{area}(\mathring{A}^2) \\ \operatorname{Residue} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Hyd}\operatorname{ropathy} \operatorname{index} \\ \operatorname{Ranking} \operatorname{of} \operatorname{amino} \operatorname{acid} \\ \operatorname{polarities} \\ \operatorname{pK}_a \\ \sigma_1 \\ \operatorname{H}_M \Delta \operatorname{PH} \\ \sigma_R \\ \sigma_\alpha \end{array}$	$ \begin{array}{r} 113. 16\\ 170\\ 166. 7\\ 124\\ 122\\ 3. 05\\ 1. 9\\ - 1. 30\\ 3\\ 9. 6\\ 0. 02\\ 0. 07\\ 0. 05\\ - 0. 04\\ \end{array} $	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$	$ \begin{array}{r} 131. 19\\ 185\\ 162. 9\\ 124\\ 137\\ 3. 43\\ 2. 3\\ - 0. 40\\ 5\\ 9. 21\\ 0. 08\\ - 0. 04\\ - 0. 12\\ - 0. 05\\ \end{array} $	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ \end{array} $	$97.12 \\ 145 \\ 112.7 \\ 90 \\ 124 \\ 3.1 \\ 1.9 \\ -3.50 \\ 13 \\ 10.6 \\ 0 \\ 0.1 \\ 0.1 \\ -0.04 \\ 10.4 \\ 10.6 \\ 10.1 \\ 0.1 \\ -0.04 \\ 10.1 \\ 10.0 \\ 10.1 \\ 10.0 \\ 10$	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ -0.05 \\ -0.02 \\ -0.02 \\ -0.02 \\ \end{array} $	$ \begin{array}{c} 140\\ 140\\ 116.1\\ 93\\ 90\\ 2.25\\ 1.1\\ 1.90\\ 12\\ 9.1\\ -0.05\\ -0.03\\ 0.02\\ -0.03\\ \end{array} $	$\begin{array}{c} 255\\ 227.8\\ 163\\ 236\\ 4.11\\ 2.9\\ 4.50\\ 6\\ 9.39\\ 0.06\\ 0.15\\ 0.09\\ -0.12\\ \end{array}$	$\begin{array}{c} 100, 103\\ 230\\ 193, 6\\ 141\\ 154\\ 2, 81\\ 1.6\\ 3, 80\\ 8\\ 9, 11\\ 0.05\\ 0.02\\ -0, 03\\ -0, 09\end{array}$	$ \begin{array}{c} 155\\ 140\\ 105\\ 135\\ 3. 38\\ 2. 2\\ - 3. 50\\ 4\\ 9. 62\\ 0. 01\\ 0. 09\\ 0. 08\\ - 0. 03\\ \end{array} $
$ \begin{array}{l} M \operatorname{ass}(\operatorname{Dalton}) \\ \operatorname{Surface} \operatorname{area}(\mathring{A}^2) \\ \operatorname{Residue} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{residue} \operatorname{non-polar} \operatorname{surface} \operatorname{area}(\mathring{A}^2) \\ \operatorname{Residue} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Hyd}\operatorname{ropathy} \operatorname{index} \\ \operatorname{Ranking} \operatorname{of} \operatorname{amino} \operatorname{acid} \\ \operatorname{polarities} \\ \operatorname{pK}_a \\ \sigma_1 \\ \operatorname{H}_M \Delta \operatorname{PH} \\ \sigma_R \\ \sigma_\alpha \\ \sigma_F \end{array} $	$ \begin{array}{r} 113. 16\\ 170\\ 166. 7\\ 124\\ 122\\ 3. 05\\ 1. 9\\ - 1. 30\\ 3\\ 9. 6\\ 0. 02\\ 0. 07\\ 0. 05\\ - 0. 04\\ - 0. 03\\ \end{array} $	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$ $0. 51$	$ \begin{array}{r} 131. 19\\ 185\\ 162. 9\\ 124\\ 137\\ 3. 43\\ 2. 3\\ - 0. 40\\ 5\\ 9. 21\\ 0. 08\\ - 0. 04\\ - 0. 12\\ - 0. 05\\ - 0. 3\\ \end{array} $	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ -0.45\\ \end{array} $	97.12 145 112.7 90 124 3.1 1.9 -3.50 13 10.6 0 0.1 0.1 -0.04 0.02	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ -0.05 \\ -0.02 \\ -0.02 \\ -0.38 \\ \end{array} $	$ \begin{array}{c} 140\\ 140\\ 116.1\\ 93\\ 90\\ 2.25\\ 1.1\\ 1.90\\ 12\\ 9.1\\ -0.05\\ -0.03\\ 0.02\\ -0.03\\ -0.44\\ \end{array} $	$\begin{array}{c} 255\\ 227.8\\ 163\\ 236\\ 4.11\\ 2.9\\ 4.50\\ 6\\ 9.39\\ 0.06\\ 0.15\\ 0.09\\ -0.12\\ -0.24\\ \end{array}$	$\begin{array}{c} 100, 103\\ 230\\ 193, 6\\ 141\\ 154\\ 2, 81\\ 1.6\\ 3, 80\\ 8\\ 9, 11\\ 0.05\\ 0.02\\ -0, 03\\ -0, 09\\ -0, 42 \end{array}$	$ \begin{array}{c} 155\\ 140\\ 105\\ 135\\ 3. 38\\ 2. 2\\ - 3. 50\\ 4\\ 9. 62\\ 0. 01\\ 0. 09\\ 0. 08\\ - 0. 03\\ - 0. 04\\ \end{array} $
$ \begin{array}{l} M \operatorname{ass}(\operatorname{Dalton}) \\ \operatorname{Surface} \operatorname{area}(\mathring{A}^2) \\ \operatorname{Residue} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{Residue} \operatorname{non-polar} \operatorname{surface} \operatorname{area}(\mathring{A}^2) \\ \operatorname{Residue} \operatorname{burial}(\operatorname{kcal} / \operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal} / \operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal} / \operatorname{mol}) \\ \operatorname{Hyd} \operatorname{ropathy} \operatorname{index} \\ \operatorname{Ranking} \operatorname{of} \operatorname{amino} \operatorname{acid} \\ \operatorname{polarities} \\ \operatorname{pK}_a \\ \sigma_1 \\ \operatorname{H}_M \Delta \operatorname{PH} \\ \sigma_R \\ \sigma_{\sigma} \\ \sigma_F \\ \operatorname{A_I} \end{array} $	$ \begin{array}{c} 113. 16\\ 170\\ 166. 7\\ 124\\ 122\\ 3. 05\\ 1. 9\\ - 1. 30\\ 3\\ 9. 6\\ 0. 02\\ 0. 07\\ 0. 05\\ - 0. 04\\ - 0. 03\\ 0. 02\\ \end{array} $	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$ $0. 51$ $0. 16$	$\begin{array}{c} 131. \ 19\\ 185\\ 162. \ 9\\ 124\\ 137\\ 3. \ 43\\ 2. \ 3\\ - \ 0. \ 40\\ 5\\ 9. \ 21\\ 0. \ 08\\ - \ 0. \ 04\\ - \ 0. \ 12\\ - \ 0. \ 05\\ - \ 0. \ 3\\ 0. \ 08\end{array}$	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ -0.45\\ 0.04\\ \end{array} $	97.12 145 112.7 90 124 3.1 1.9 -3.50 13 10.6 0 0.1 0.1 -0.04 0.02 0	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ -0.05 \\ -0.02 \\ -0.02 \\ -0.38 \\ 0.03 \\ \end{array} $	$ \begin{array}{c} 100, 11\\ 140\\ 116, 1\\ 93\\ 90\\ 2, 25\\ 1, 1\\ 1, 90\\ 12\\ 9, 1\\ -0, 05\\ -0, 03\\ 0, 02\\ -0, 03\\ -0, 44\\ 0, 05\\ \end{array} $	$\begin{array}{c} 255\\ 255\\ 227.8\\ 163\\ 236\\ 4.11\\ 2.9\\ 4.50\\ 6\\ 9.39\\ 0.06\\ 0.15\\ 0.09\\ -0.12\\ -0.24\\ 0.06\\ \end{array}$	$ \begin{array}{c} 103.13\\ 230\\ 193.6\\ 141\\ 154\\ 2.81\\ 1.6\\ 3.80\\ 8\\ 9.11\\ 0.05\\ 0.02\\ -0.03\\ -0.09\\ -0.42\\ 0.05\\ \end{array} $	$ \begin{array}{c} 155\\ 140\\ 105\\ 135\\ 3. 38\\ 2. 2\\ - 3. 50\\ 4\\ 9. 62\\ 0. 01\\ 0. 09\\ 0. 08\\ - 0. 03\\ - 0. 04\\ 0. 01\\ \end{array} $
$ \begin{array}{l} M \operatorname{ass}(\operatorname{Dalton}) \\ \operatorname{Surface} \operatorname{area}(\mathring{A}^2) \\ \operatorname{Residue} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\mathring{A}^3) \\ \operatorname{Residue} \operatorname{non-polar} \operatorname{surface} \operatorname{area}(\mathring{A}^2) \\ \operatorname{Residue} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal}/\operatorname{mol}) \\ \operatorname{Hyd}\operatorname{ropathy} \operatorname{index} \\ \operatorname{Ranking} \operatorname{of} \operatorname{amino} \operatorname{acid} \\ \operatorname{polarities} \\ \operatorname{pK}_a \\ \sigma_I \\ \operatorname{H}_M \Delta \operatorname{PH} \\ \sigma_R \\ \sigma_a \\ \sigma_F \\ \operatorname{A}_I \\ \operatorname{P(a)} \\ \end{array} \right) $	$\begin{array}{c} 113. 16 \\ 170 \\ 166. 7 \\ 124 \\ 122 \\ 3. 05 \\ 1. 9 \\ -1. 30 \\ 3 \\ 9. 6 \\ 0. 02 \\ 0. 07 \\ 0. 05 \\ -0. 04 \\ -0. 03 \\ 0. 02 \\ 121 \end{array}$	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$ $0. 51$ $0. 16$ 114	$ \begin{array}{c} 131. 19\\ 185\\ 162. 9\\ 124\\ 137\\ 3. 43\\ 2. 3\\ - 0. 40\\ 5\\ 9. 21\\ 0. 08\\ - 0. 04\\ - 0. 12\\ - 0. 05\\ - 0. 3\\ 0. 08\\ 145\\ \end{array} $	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ -0.45\\ 0.04\\ 113\\ \end{array} $	97.12 145 112.7 90 124 3.1 1.9 -3.50 13 10.6 0 0.1 0.1 -0.04 0.02 0 57	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ -0.05 \\ -0.02 \\ -0.02 \\ -0.38 \\ 0.03 \\ 77 \\ \end{array} $	$ \begin{array}{c} 100, 11\\ 140\\ 116, 1\\ 93\\ 90\\ 2, 25\\ 1, 1\\ 1, 90\\ 12\\ 9, 1\\ -0, 05\\ -0, 03\\ 0, 02\\ -0, 03\\ -0, 44\\ 0, 05\\ 83\\ \end{array} $	$\begin{array}{c} 255\\ 227.8\\ 163\\ 236\\ 4.11\\ 2.9\\ 4.50\\ 6\\ 9.39\\ 0.06\\ 0.15\\ 0.09\\ -0.12\\ -0.24\\ 0.06\\ 108\\ \end{array}$	$\begin{array}{c} 100, 103\\ 230\\ 193, 6\\ 141\\ 154\\ 2, 81\\ 1.6\\ 3.80\\ 8\\ 9, 11\\ 0.05\\ 0.02\\ -0.03\\ -0.09\\ -0.42\\ 0.05\\ 69\end{array}$	$ \begin{array}{c} 155\\ 140\\ 105\\ 135\\ 3. 38\\ 2. 2\\ - 3. 50\\ 4\\ 9. 62\\ 0. 01\\ 0. 09\\ 0. 08\\ - 0. 03\\ - 0. 04\\ 0. 01\\ 106\\ \end{array} $
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial (kcal/ mol) Hydropathy index Ranking of amino acid polarities pK_a σ_I $H_M \Delta PH$ σ_R σ_a σ_F A_I P(a) P(b)	$\begin{array}{c} 113. 16 \\ 170 \\ 166. 7 \\ 124 \\ 122 \\ 3. 05 \\ 1. 9 \\ -1. 30 \\ 3 \\ 9. 6 \\ 0. 02 \\ 0. 07 \\ 0. 05 \\ -0. 04 \\ -0. 03 \\ 0. 02 \\ 121 \\ 130 \\ \end{array}$	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$ $0. 51$ $0. 16$ 114 74	$\begin{array}{c} 131. \ 19\\ 185\\ 162. \ 9\\ 124\\ 137\\ 3. \ 43\\ 2. \ 3\\ - \ 0. \ 40\\ 5\\ 9. \ 21\\ 0. \ 08\\ - \ 0. \ 04\\ - \ 0. \ 12\\ - \ 0. \ 05\\ - \ 0. \ 3\\ 0. \ 08\\ 145\\ 105\\ \end{array}$	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ -0.45\\ 0.04\\ 113\\ 138 \end{array} $	97.12 145 112.7 90 124 3.1 1.9 -3.50 13 10.6 0 0.1 0.1 -0.04 0.02 0 57 55	87.08 115 89 73 56 1.4 0.2 2.80 14 9.15 -0.03 -0.05 -0.02 -0.38 0.03 77 75	$ \begin{array}{c} 101, 11\\ 140\\ 116, 1\\ 93\\ 90\\ 2, 25\\ 1, 1\\ 1, 90\\ 12\\ 9, 1\\ -0, 05\\ -0, 03\\ 0, 02\\ -0, 03\\ -0, 44\\ 0, 05\\ 83\\ 119\\ \end{array} $	$\begin{array}{c} 255\\ 227.8\\ 163\\ 236\\ 4.11\\ 2.9\\ 4.50\\ 6\\ 9.39\\ 0.06\\ 0.15\\ 0.09\\ -0.12\\ -0.24\\ 0.06\\ 108\\ 137\\ \end{array}$	$\begin{array}{c} 100, 103\\ 230\\ 193, 6\\ 141\\ 154\\ 2, 81\\ 1.6\\ 3.80\\ 8\\ 9, 11\\ 0.05\\ 0.02\\ -0.03\\ -0.09\\ -0.42\\ 0.05\\ 69\\ 147\\ \end{array}$	$ \begin{array}{c} 155\\140\\105\\135\\3.38\\2.2\\-3.50\\4\\9.62\\0.01\\0.09\\0.08\\-0.03\\-0.04\\0.01\\106\\170\end{array} $
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial (kcal/ mol) Hydropathy index Ranking of amino acid polarities pK_a σ_I $H_M \Delta PH$ σ_R σ_{α} σ_F A _I P(a) P(b) P(turn)	$\begin{array}{c} 113. 16 \\ 170 \\ 166. 7 \\ 124 \\ 122 \\ 3. 05 \\ 1. 9 \\ -1. 30 \\ 3 \\ 9. 6 \\ 0. 02 \\ 0. 07 \\ 0. 05 \\ -0. 04 \\ -0. 03 \\ 0. 02 \\ 121 \\ 130 \\ 59 \end{array}$	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$ $0. 51$ $0. 16$ 114 74 101	$\begin{array}{c} 131. \ 19\\ 185\\ 162. \ 9\\ 124\\ 137\\ 3. \ 43\\ 2. \ 3\\ - \ 0. \ 40\\ 5\\ 9. \ 21\\ 0. \ 08\\ - \ 0. \ 04\\ - \ 0. \ 12\\ - \ 0. \ 05\\ - \ 0. \ 3\\ 0. \ 08\\ 145\\ 105\\ 60\\ \end{array}$	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ -0.45\\ 0.04\\ 113\\ 138\\ 60\\ \end{array} $	97.12 145 112.7 90 124 3.1 1.9 -3.50 13 10.6 0 0.1 0.1 -0.04 0.02 0 57 55 152	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ -0.05 \\ -0.02 \\ -0.38 \\ 0.03 \\ 77 \\ 75 \\ 143 \\ \end{array} $	$ \begin{array}{c} 140\\ 140\\ 116.1\\ 93\\ 90\\ 2.25\\ 1.1\\ 1.90\\ 12\\ 9.1\\ -0.05\\ -0.03\\ 0.02\\ -0.03\\ -0.44\\ 0.05\\ 83\\ 119\\ 96\\ \end{array} $	$\begin{array}{c} 255\\ 227.8\\ 163\\ 236\\ 4.11\\ 2.9\\ 4.50\\ 6\\ 9.39\\ 0.06\\ 0.15\\ 0.09\\ -0.12\\ -0.24\\ 0.06\\ 108\\ 137\\ 96\end{array}$	$\begin{array}{c} 100, 103\\ 230\\ 193, 6\\ 141\\ 154\\ 2, 81\\ 1.6\\ 3.80\\ 8\\ 9, 11\\ 0.05\\ 0.02\\ -0.03\\ -0.09\\ -0.42\\ 0.05\\ 69\\ 147\\ 114\\ \end{array}$	$ \begin{array}{c} 155\\ 140\\ 105\\ 135\\ 3.38\\ 2.2\\ -3.50\\ 4\\ 9.62\\ 0.01\\ 0.09\\ 0.08\\ -0.03\\ -0.04\\ 0.01\\ 106\\ 170\\ 50\\ \end{array} $
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial (kcal/ mol) Hydropathy index Ranking of amino acid polarities pK_a σ_I $H_M \Delta PH$ σ_R σ_{α} σ_F A _I P(a) P(b) P(turn) f(j)	$\begin{array}{c} 113. 16 \\ 170 \\ 166. 7 \\ 124 \\ 122 \\ 3. 05 \\ 1. 9 \\ -1. 30 \\ 3 \\ 9. 6 \\ 0. 02 \\ 0. 07 \\ 0. 05 \\ -0. 04 \\ -0. 03 \\ 0. 02 \\ 121 \\ 130 \\ 59 \\ 0. 061 \end{array}$	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$ $0. 51$ $0. 16$ 114 74 101 $0. 055$	$\begin{array}{c} 131. \ 19\\ 185\\ 162. \ 9\\ 124\\ 137\\ 3. \ 43\\ 2. \ 3\\ - \ 0. \ 40\\ 5\\ 9. \ 21\\ 0. \ 08\\ - \ 0. \ 04\\ - \ 0. \ 12\\ - \ 0. \ 05\\ - \ 0. \ 3\\ 0. \ 08\\ 145\\ 105\\ 60\\ 0. \ 068\\ \end{array}$	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ -0.45\\ 0.04\\ 113\\ 138\\ 60\\ 0.059\\ \end{array} $	97.12 145 112.7 90 124 3.1 1.9 -3.50 13 10.6 0 0.1 0.1 -0.04 0.02 0 57 55 152 0.102	87.08 115 89 73 56 1.4 0.2 2.80 14 9.15 -0.03 -0.05 -0.02 -0.38 0.03 77 75 143 0.12	$ \begin{array}{c} 140\\ 140\\ 140\\ 116.1\\ 93\\ 90\\ 2.25\\ 1.1\\ 1.90\\ 12\\ 9.1\\ -0.05\\ -0.03\\ 0.02\\ -0.03\\ -0.44\\ 0.05\\ 83\\ 119\\ 96\\ 0.086\\ \end{array} $	$\begin{array}{c} 255\\ 227.8\\ 163\\ 236\\ 4.11\\ 2.9\\ 4.50\\ 6\\ 9.39\\ 0.06\\ 0.15\\ 0.09\\ -0.12\\ -0.24\\ 0.06\\ 108\\ 137\\ 96\\ 0.077\\ \end{array}$	$\begin{array}{c} 100, 103\\ 230\\ 193, 6\\ 141\\ 154\\ 2, 81\\ 1.6\\ 3.80\\ 8\\ 9, 11\\ 0.05\\ 0.02\\ -0, 03\\ -0, 09\\ -0, 42\\ 0.05\\ 69\\ 147\\ 114\\ 0, 082 \end{array}$	$ \begin{array}{c} 155\\ 140\\ 105\\ 135\\ 3. 38\\ 2. 2\\ -3. 50\\ 4\\ 9. 62\\ 0. 01\\ 0. 09\\ 0. 08\\ -0. 03\\ -0. 04\\ 0. 01\\ 106\\ 170\\ 50\\ 0. 062\\ \end{array} $
Mass(Dalton) Surface area(Å ²) Residue volume(Å ³) van der Waals volume (Å ³) Residue non-polar sur- face area(Å ²) Residue burial (kcal/ mol) Side chain burial (kcal/ mol) Hydropathy index Ranking of amino acid polarities pK_a σ_I $H_M \Delta PH$ σ_R σ_{α} σ_F A ₁ P(a) P(b) P(turn) f(i) f(i+1)	$\begin{array}{c} 113. 16 \\ 170 \\ 166. 7 \\ 124 \\ 122 \\ 3. 05 \\ 1. 9 \\ -1. 30 \\ 3 \\ 9. 6 \\ 0. 02 \\ 0. 07 \\ 0. 05 \\ -0. 04 \\ -0. 03 \\ 0. 02 \\ 121 \\ 130 \\ 59 \\ 0. 061 \\ 0. 025 \end{array}$	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$ $0. 51$ $0. 16$ 114 74 101 $0. 055$ $0. 115$	$\begin{array}{c} 131. \ 19\\ 185\\ 162. \ 9\\ 124\\ 137\\ 3. \ 43\\ 2. \ 3\\ - \ 0. \ 40\\ 5\\ 9. \ 21\\ 0. \ 08\\ - \ 0. \ 04\\ - \ 0. \ 12\\ - \ 0. \ 05\\ - \ 0. \ 3\\ 0. \ 08\\ 145\\ 105\\ 60\\ 0. \ 068\\ 0. \ 082\\ \end{array}$	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ -0.45\\ 0.04\\ 113\\ 138\\ 60\\ 0.059\\ 0.041 \end{array} $	$\begin{array}{c} 97.12 \\ 145 \\ 112.7 \\ 90 \\ 124 \\ 3.1 \\ 1.9 \\ -3.50 \\ 13 \\ 10.6 \\ 0 \\ 0.1 \\ 0.1 \\ -0.04 \\ 0.02 \\ 0 \\ 57 \\ 55 \\ 152 \\ 0.102 \\ 0.301 \end{array}$	$ \begin{array}{r} 87.08 \\ 115 \\ 89 \\ 73 \\ 56 \\ 1.4 \\ 0.2 \\ 2.80 \\ 14 \\ 9.15 \\ -0.03 \\ -0.05 \\ -0.02 \\ -0.38 \\ 0.03 \\ 77 \\ 75 \\ 143 \\ 0.12 \\ 0.139 \\ \end{array} $	$\begin{array}{c} 140\\ 116. 1\\ 93\\ 90\\ 2. 25\\ 1. 1\\ 1. 90\\ 12\\ 9. 1\\ -0. 05\\ -0. 03\\ 0. 02\\ -0. 03\\ -0. 44\\ 0. 05\\ 83\\ 119\\ 96\\ 0. 086\\ 0. 108\\ \end{array}$	$\begin{array}{c} 255\\ 227.\ 8\\ 163\\ 236\\ 4.\ 11\\ 2.\ 9\\ 4.\ 50\\ 6\\ 9.\ 39\\ 0.\ 06\\ 0.\ 15\\ 0.\ 09\\ -0.\ 12\\ -0.\ 24\\ 0.\ 06\\ 108\\ 137\\ 96\\ 0.\ 077\\ 0.\ 013\\ \end{array}$	$\begin{array}{c} 100.16\\ 230\\ 193.6\\ 141\\ 154\\ 2.81\\ 1.6\\ 3.80\\ 8\\ 9.11\\ 0.05\\ 0.02\\ -0.03\\ -0.09\\ -0.42\\ 0.05\\ 69\\ 147\\ 114\\ 0.082\\ 0.065\\ \end{array}$	$ \begin{array}{r} 155 \\ 140 \\ 105 \\ 135 \\ 3. 38 \\ 2. 2 \\ - 3. 50 \\ 4 \\ 9. 62 \\ 0. 01 \\ 0. 09 \\ 0. 08 \\ - 0. 03 \\ - 0. 04 \\ 0. 01 \\ 106 \\ 170 \\ 50 \\ 0. 062 \\ 0. 048 \\ \end{array} $
$ \begin{array}{l} M \operatorname{ass}(\operatorname{Dalton}) \\ \operatorname{Surface} \operatorname{area}(\mathbb{A}^2) \\ \operatorname{Residue} \operatorname{volume}(\mathbb{A}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\mathbb{A}^3) \\ \operatorname{van} \operatorname{der} W \operatorname{aals} \operatorname{volume}(\mathbb{A}^3) \\ \operatorname{Residue} \operatorname{non-polar} \operatorname{surface} \operatorname{area}(\mathbb{A}^2) \\ \operatorname{Residue} \operatorname{burial}(\operatorname{kcal} / \operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal} / \operatorname{mol}) \\ \operatorname{Side} \operatorname{chain} \operatorname{burial}(\operatorname{kcal} / \operatorname{mol}) \\ \operatorname{Hyd} \operatorname{ropathy} \operatorname{index} \\ \operatorname{Ranking} \operatorname{of} \operatorname{amino} \operatorname{acid} \\ \operatorname{polarities} \\ \operatorname{pK}_a \\ \sigma_I \\ \operatorname{H}_M \Delta P \\ \\ \sigma_R \\ \sigma_\alpha \\ \sigma_F \\ A_I \\ P(a) \\ P(b) \\ P(turn) \\ f(i) \\ f(i+1) \\ f(i+2) \\ \end{array} $	$\begin{array}{c} 113. 16 \\ 170 \\ 166. 7 \\ 124 \\ 122 \\ 3. 05 \\ 1. 9 \\ -1. 30 \\ 3 \\ 9. 6 \\ 0. 02 \\ 0. 07 \\ 0. 05 \\ -0. 04 \\ -0. 03 \\ 0. 02 \\ 121 \\ 130 \\ 59 \\ 0. 061 \\ 0. 025 \\ 0. 036 \\ \end{array}$	128. 17 200 $168. 6$ 135 164 $4. 1$ $2. 9$ $2. 50$ 20 $8. 95$ $- 0. 16$ $- 1. 11$ $- 0. 95$ $- 0. 05$ $0. 51$ $0. 16$ 114 74 101 $0. 055$ $0. 115$ $0. 072$	$\begin{array}{c} 131. \ 19\\ 185\\ 162. \ 9\\ 124\\ 137\\ 3. \ 43\\ 2. \ 3\\ - \ 0. \ 40\\ 5\\ 9. \ 21\\ 0. \ 08\\ - \ 0. \ 04\\ - \ 0. \ 12\\ - \ 0. \ 05\\ - \ 0. \ 3\\ 0. \ 08\\ 145\\ 105\\ 60\\ 0. \ 068\\ 0. \ 082\\ 0. \ 014\\ \end{array}$	$ \begin{array}{r} 147.18\\ 210\\ 189.9\\ 135\\ 194\\ 3.46\\ 2.3\\ -3.20\\ 2\\ 9.13\\ 0.04\\ 0.06\\ 0.02\\ -0.08\\ -0.45\\ 0.04\\ 113\\ 138\\ 60\\ 0.059\\ 0.041\\ 0.065 \end{array} $	$\begin{array}{c} 97.12 \\ 145 \\ 112.7 \\ 90 \\ 124 \\ 3.1 \\ 1.9 \\ -3.50 \\ 13 \\ 10.6 \\ 0 \\ 0.1 \\ 0.1 \\ -0.04 \\ 0.02 \\ 0 \\ 57 \\ 55 \\ 152 \\ 0.102 \\ 0.301 \\ 0.034 \end{array}$	$\begin{array}{c} 87.08\\ 115\\ 89\\ 73\\ 56\\ 1.4\\ 0.2\\ 2.80\\ 14\\ 9.15\\ -0.03\\ -0.05\\ -0.02\\ -0.02\\ -0.38\\ 0.03\\ 77\\ 75\\ 143\\ 0.12\\ 0.139\\ 0.125\\ \end{array}$	$\begin{array}{c} 140\\ 116. 1\\ 93\\ 90\\ 2. 25\\ 1. 1\\ 1. 90\\ 12\\ 9. 1\\ -0. 05\\ -0. 03\\ 0. 02\\ -0. 03\\ -0. 44\\ 0. 05\\ 83\\ 119\\ 96\\ 0. 086\\ 0. 108\\ 0. 065\\ \end{array}$	$\begin{array}{c} 255\\ 227.\ 8\\ 163\\ 236\\ 4.\ 11\\ 2.\ 9\\ 4.\ 50\\ 6\\ 9.\ 39\\ 0.\ 06\\ 0.\ 15\\ 0.\ 09\\ -0.\ 12\\ -0.\ 24\\ 0.\ 06\\ 108\\ 137\\ 96\\ 0.\ 077\\ 0.\ 013\\ 0.\ 064 \end{array}$	$\begin{array}{c} 100.163\\ 230\\ 193.6\\ 141\\ 154\\ 2.81\\ 1.6\\ 3.80\\ 8\\ 9.11\\ 0.05\\ 0.02\\ -0.03\\ -0.09\\ -0.42\\ 0.05\\ 69\\ 147\\ 114\\ 0.082\\ 0.065\\ 0.114\\ \end{array}$	$ \begin{array}{r} 155 \\ 140 \\ 105 \\ 135 \\ 3. 38 \\ 2. 2 \\ - 3. 50 \\ 4 \\ 9. 62 \\ 0. 01 \\ 0. 09 \\ 0. 08 \\ - 0. 03 \\ - 0. 04 \\ 0. 01 \\ 106 \\ 170 \\ 50 \\ 0. 062 \\ 0. 048 \\ 0. 028 \\ \end{array} $

 σ_1 : Inductive effect scale $H_M \Delta PH$: Normalized Mulliken population data for the amino-acid side chains in the context of phenol, σ_R : Resonance effect scale σ_a : Normalized polarizability index, σ_F : Field effect index, A_1 : Additional scale f(i): Frequency of the 1st residue in turn, f(i+1): Frequency of the 2nd residue in turn, f(i+2): Frequency of the 3rd residue in turn, f(i+3): Frequency of the 4th residue in turn.

	Sharoonaase.	
Amino acid	Number	Distribution probability
А	54	0.015
R	19	0.017
Ν	21	0. 027
D	42	8.903e-3
С	6	0.039
Е	26	0.040
Q	23	0.040
G	35	5. 699e-3
Н	10	0. 191
Ι	32	0. 037
L	55	0.012
K	23	7. 082e-4
М	14	0. 055
F	24	0.040
Р	26	4.712e-5
S	49	1. 650e-5
Т	48	8. 611e-3
W	12	4. 432e-3
Y	22	0. 051
V	31	9. 435e-3

Table 2 Amino- acid composition and distribution probabilities of O9 AT27 β -glucosidase

1.4 Model development

Of 39 β -glucosidases listed in Table 3, 23 served as training group to generate the neural network model parameters, weights and biases, and 16 served as validation group. This is a very traditional approach.

A more recent approach is the jackknife, and the jackknifing of delete-1 observation was used¹⁹, i.e. one β -glucosidase of 39 β -glucosidases did not attend the training, while the generated model parameters were used to predict optimum pH value and optimum temperature of omitted β -glucosidase, until that each β -glucosidase went through this jackknifing process.

The third approach is the cross-validation, i.e. 39β -glucosidases were split into 3 subsets containing 13β -glucosidases each, then one subset did not attend the training, but the generated model parameters were used to predict optimum pH value and optimum temperature in omitted subset, until that each subset went through this cross-validation process.

A nother way to divide 39 β -glucosidases was to split β -glucosidases into 13 subsets containing 3 β -glucosidases each, and to go to the same process as described above.

The above three approaches were applied to each predictor listed in Table 1 in order to compare their predictions statistically.

1.5 Statistics

For each predictor, one hundred trainings were conducted, and the obtained 100 sets of weights and biases were used to predict optimum pH value and optimum temperature 100 times, and their mean and standard deviation were used to compare the recorded optimum pH value and optimum temperature for each β -glucosidase^[20].

2 Results

Fig. 1 is the scheme of neural network for model development. This model is particularly designed to simultaneously account for the features of aminoacids and optimum pH value and optimum temperature of β -glucosidases.

For training of neural network, the initialization of weights and biases and number of training epochs govern whether the neural network can converge. We used the random initialization function to initialize weights and biases, and 350 training epochs. Fig. 2 displays training processes in 23 β -glucosidases with different features of amino acids (Table 1). In this figure, each line represents a training process from the beginning to the end, and we can find the convergence reached within 350 training epochs with any random initialization, which lays the foundation to guarantee our training process.

In Fig. 3, we can see that the percentage of correctly predicted β -glucosidases improves with respect to training epochs. Actually, what we need to see is whether this percentage is stable along the training process, which is the case in Fig. 3, so we can exclude the possibility of over-fitting or overgeneralization using this neural network model. Also, Fig.3 demonstrates that predictions of optimum pH value and optimum temperature using some features of amino acids can reach a pretty good level. As we used three different approaches to develop predictive models, Fig. 3 is only related to one approach, and Fig. 4 shows the correctly predicted percentage improves with respect to training epochs in the other two approaches. In general, all three approaches reach the similar results.



Fig. 1 20-2 feedforward backpropagation neural network to model the relationship between 20 amino-acid features of β -glucosidase and optimum pH value and optimum temperature. Each tri-circle represents a neuron.

 $IW\{1\}$: the input weights $LW\{2, 1\}$: the layer weights to the second layer from the first layer, $b\{1\}$ and $b\{2\}$: the biases related to each neuron at the first and second layers.



Fig. 2 Convergence of mean squared error performance function with 100 different initial weights and biases generated by random initialization function in training

 $(a)\sigma_1 \times No., (b)\sigma_{\alpha} \times No., (c)f(i) \times No., (d)DP.$

No.: amino-acid composition σ_1 : inductive effect scale, σ_{α} : normalized polarizability index, f(i): frequency of the 1st residue in turn DP: amino-acid distribution probability.



Fig. 3 Percentage of correctly predicted optimum pH value and optimum temperature by different features of amino acids

(a) Training, (b) Validation, (c) Total.

The training and validation groups contain 23 and 16 β -glucosidases.

•••: pH by $\sigma_1 \times No.$, -••: pH by $\sigma_{\alpha} \times No.$, -••: pH by f (i)×No., -••: pH by DP, -••: Tm by $\sigma_1 \times No.$, -••: Tm by $\sigma_{\alpha} \times No.$, -••: Tm by f(i)×No., -••: Tm by DP.

pH: optimum pH value, Tm: optimum temperature, No.: amino-acid composition, σ_1 : inductive effect scale, σ_α : normalized polarizability index, f(i): frequency of the 1st residue in turn, DP: amino-acid distribution probability, \times : multiplication.

Table 3 details the statistical comparison between predicted and recorded optimum pH value and optimum temperature. The data generated in Table 3 are based on the very traditional approach to divide the training and validation groups, where the prediction should be good if there is no statistical difference between recorded and predicted optimum pH value and optimum temperature, respectively. The prediction based on amino-acid distribution probability is clearly better than the predictions based on other features of amino acids.

Table 3 Statistical comparison between recorded and predicted optimum pH value and opti	mum temperature(mean $\pm SD$, $n=100$)
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	cœssion	Recorded	Optimum pH value predicted by predictor			Becorded	d Optimum temperature predicted by predictor				
Group	number	pH – value	$\sigma_I \times No.$	$\sigma_{\alpha} \times No.$	f(i) x No.	DP	Tm	$\sigma_I \!\! \times \! N$ o.	$\sigma_{\alpha}\!\!\times No.$	f(i)×No.	DP
Train-	Q9A T27	4	5. 83 ±0. 10	5.86±0.14	5. 83 ±0. 09	4.05±0.13*	50	46.39±4.92 *	44. 64±5. 73 *	47. 43 ±4. 24 *	50.00 \pm 0.02 *
Ing	Q8TGI8	4	5. 70±0. 21	5.57 \pm 0.32	5.73 \pm 0.18	4.35±0.35 *	71.5	52.80±6.86	56.58±8.82 *	51.74±6.40	71.48 \pm 0.07 *
	Q12715	4.6	5. 64 \pm 0. 28	5. 54 \pm 0. 34	5. 71 ±0. 21	4.66±0.25 *	70	56.34±11.80*	58.90±11.34	* 52.66±8.53	69. 91 ± 0 . 76 *
	A1C3J9	5	5. 82 \pm 0. 08	5. 85 \pm 0. 09	5. 81 \pm 0. 07	5.00±0.13 *	40	47.47±3.05	45.71±4.08 *	47. 98±2. 97	40.00 \pm 0.08 *
	Q8T0W7	5	5.75 \pm 0.11	5.77 \pm 0.13	5.78 \pm 0.07	5.07±0.23 *	50	$51.94{\pm}5.17$ *	51. 19 \pm 3. 35 *	50. 02 \pm 2. 70 *	50. 10 \pm 0. 78 *
	Q4U4W7	5	5.74 \pm 0.14	5. 55 $\pm 0.$ 35 $ *$	5. 71 ±0. 21	5.03±0.12*	50	$50.77 {\pm} 5.17 *$	57. 29 \pm 10. 90	* 52. 13±7. 98 *	50.01 ± 0.03 *
	A9UIG0	5	5.66 \pm 0.26	5.54 ± 0.34 *	5. 71 ±0. 21	4.93±0.25 *	70	55.32±10.19*	58.71±11.27	* 52. 69±8. 67 *	70.02 ± 0.10 *
	P94248	5.5	5. 85 \pm 0. 11	5.90 \pm 0.16	5.83 \pm 0.10	5.51±0.11*	45	$45.80{\pm}5.43$ *	43. 37±6. 56 *	47.26 \pm 4.54 *	45.00 ± 0.04 *
	008331	5.5	5.73 ± 0.16 *	5.62 ± 0.26 *	5.73 \pm 0.16 $*$	5.51±0.15 *	65	53.63±7.55 *	56. 21±8. 20 *	52. 03±6. 73 *	65.00 ± 0.02 *
	Q9SLA0	5.6	5.84±0.11	5. 89±0. 13	5. 83 ±0. 09	5.64±0.12*	40	46.14±5.19 *	44. 31±5. 44 *	47. 43±4. 12 *	40.00 \pm 0.02 *
	P49235	5.8	5. 83 $\pm 0.$ 09 *	5. 83 $\pm 0.$ 11 *	5. 81 \pm 0. 06 *	5.67±0.19*	50	46.56±4.56 *	46. 12±4. 46 *	48. 19±3. 03 *	50.00 \pm 0.04 *
	Q86D78	6	5. 85 $\pm 0.$ 11 *	5. 91 $\pm 0.$ 16 *	5. 83 \pm 0. 10 *	6.04±0.12*	35	45.69±5.60 *	43. 31±6. 55 *	47 . 28 ±4 . 48	35.00 \pm 0.03 *
	Q2WGB4	46	5. 83 \pm 0. 08	5. 88 $\pm 0.$ 12 *	5. 82±0. 09	5.99±0.09*	37	46.94±3.78	44.76±5.01 *	47.65±3.77	37.00 \pm 0.03 *
	Q875K3	6	5. 82 $\pm 0.$ 11 *	5. 87 $\pm 0.$ 12 *	5. 83 ±0. 09 *	5.98±0.09*	40	46.73±5.47 *	44. 24±5. 57 *	47.38±4.26 *	40.00 \pm 0.02 *
	Q25BW5	6.5	5. 86 \pm 0. 15	5. 91 \pm 0. 18	5. 83 ±0. 09	6.47±0.13 *	30	45.91±5.47	44. 39±5. 90	47. 45±4. 09	30. 01 $\pm 0.$ 05 *
	P15885	6.5	5. 85 \pm 0. 12	5. 91 ±0. 16	5.83±0.10	6.46±0.11*	30	45.68±5.49	43. 18±6. 75 *	47.26±4.52	30.00 \pm 0.03 *
	Q59976	6.5	5.86±0.19	5. 91 ±0. 19	5. 83 ±0. 09	6.54±0.11*	30	46.03±6.29	43. 89±7. 00 *	47.45±4.09	30.00 \pm 0.03 *
	Q9H227	6.5	5.84±0.09	5.88±0.14	5.83±0.10	6.43±0.12*	50	46.08±4.95 *	44.48±5.64 *	47.36±4.30 *	50.01 ± 0.04 *
	Q746L1	6.5	5. 70±0. 24	5.68±0.34	5. 73 ±0. 21	6.44±0.21*	88	57.95±15.12*	62.97±16.16	* 53. 58±10.37	87. 99 \pm 0. 04 *
	B9K7M5	6.5	5.68±0.25	5.68±0.37	5. 73 ±0. 21	6.35±0.21*	95	58.47±14.94	64.33±17.67	* 53. 58±10.59	94. 96 \pm 0. 18 *
	Q08IT 7	7	5.85±0.13	5. 89±0. 15	5. 82 \pm 0. 08	6.93±0.12*	30	46.20±5.24	44. 36±5. 71	47. 74±3. 61	30.00 \pm 0.03 *
	Q6QGY5	7	5.85±0.11	5.89±0.14	5.83±0.10	6.98±0.11*	40	45.91±5.09 *	43. 97±5. 75 *	47. 29±4. 46 *	40. 01 \pm 0. 04 *
	Q47RE2	7.2	5.85 \pm 0.15	5.90±0.17	5. 83 ±0. 09	7.14±0.13 *	25	46.08±5.94	43. 87±6. 77	47. 39±4. 22	24. 99 \pm 0. 06 *
V alid-	Q08638	3.2	5. 69±0. 24	5.67±0.37	5.74±0.21	5.71±1.26*	85	58.43±15.30°	64.99±18.80	* 53.66±10.85	67. 71 ±14.73 *
atio n	B5TWK3	4.5	5. 62±0. 31	5. 51 ±0. 36	5. 71 ±0. 22	5.12±0.90*	22	58.70±17.26	61.04±16.14	52. 92±9. 54	59.08±9.07
	Q12715	4.6	5. 64 \pm 0. 28	5. 54 \pm 0. 34	5. 71 ±0. 21	4.66±0.25 *	65	56.34±11.80°	58.90±11.34	* 52. 66±8. 53 *	69. 91 ± 0 . 76
	B5TWK3	5	5. 62 \pm 0. 31	5. 51 $\pm 0.$ 36 *	5. 71 ±0. 22	5.12±0.90*	37	58.70±17.26*	61.04±16.14	* 52. 92±9. 54 *	59.08 \pm 9.07
	B6ZKM3	5	5.81±0.14	5.79 \pm 0.27	5.78 \pm 0.09	5.92±1.59*	30	48.50±5.77	51.01 ± 10.65	* 49 . 31 ±3 . 72	49. 29±17.14 *
	Q2UUD6	5 5	5.68 \pm 0.20	5. 57 $\pm 0.$ 32 *	5.72 \pm 0.18	4.96±0.66*	60	54.37±10.84*	56.98±11.75	* 51. 92 ±7. 17 *	60. 20 \pm 4. 11 *
	Q9SPK3	5	5.84±0.11	5.90 \pm 0.15	5. 83 ±0. 09	6.09±0.98 *	50	46.03 \pm 5.38 *	43.75±6.48 *	47. 32 \pm 4. 38 *	62. 34 \pm 9. 92 *
	A9UIG0	6	5.66 ± 0.26 *	5. 54 $\pm 0.$ 34 *	5.71±0.21 *	$4.93{\pm}0.25$	70	55.32±10.19*	58.71±11.27	* 52. 69±8. 67 *	70.02 ± 0.10 *
	061594	6	5. 83 $\pm 0.$ 09 *	5.87 ± 0.11 *	5. 82 \pm 0. 08	5.53±1.29*	30	46.64±4.28	44 . 75±5. 14	47.88±3.89	60.57 \pm 10.65
	Q12601	6	5. 80 $\pm 0.$ 19 *	5.78 ± 0.25 *	5.79 \pm 0.12 *	5.26±1.00 *	35	$48.04{\pm}5.40$	48.37±8.17 *	48. 83±4.06	60.97 \pm 10.07
	P26208	6	5. 85 $\pm 0.$ 18 *	5.89 \pm 0.26 *	5. 81 \pm 0. 10 *	6.84±1.22 *	65	47.14±6.83	48.22±8.77 *	48. 77±4. 16	59. 84±14.00 *
	P10482	6	5. 80 \pm 0. 13 *	5.79 ± 0.16 *	5.79 \pm 0.05	5.54±1.55*	80	49.34±5.86	50. 58 \pm 5. 71	49 . 24 ±2 . 64	57.36±17.67*
	P96316	6.2	5.68 \pm 0.23	5.57 ± 0.34 *	5. 71 \pm 0. 20	5.76±1.23 *	35	55.81±14.34 *	61.65±16.69	* 53. 46±12.27	* 65. 98±11.55
	Q9C3Z9	6.4	5. 70±0. 20	5. 59±0. 32	5. 71 ±0. 20	5.36±1.05 *	50	54.70±12.02 *	60. 20±16.63	* 53. 20±10.99	* 73. 78±13.01 *
	Q9H227	6.5	5.84±0.10	5.88±0.14	5. 83 ±0. 10	6.51±0.30*	50	46.03±4.97 *	44. 40±5. 66 *	47. 35±4. 32 *	47.99±2.88 *
	P96316	6.6	5.68±0.23	5. 57±0. 34	5. 71 ±0. 20	5.76±1.23*	45	55.81±14.34*	61.65±16.69	* 53. 46 ± 12. 27	* 65. 98±11.55 *
Total perfor- mance	_	—	9	15	7	38	—	24	33	20	33

No. ; am ino-acid composition, σ_1 : inductive effect scale, σ_{α} : normalized polarizability index, f(i): frequency of the 1st residue in turn DP: am ino-acid distribution probability, Tm; temperature, \times : multiplication.

3 Discussion

The model used in this study can account for any possible interaction between pH value and tem-

perature if such an interaction would exist. Statistically, the two-way ANOVA could detect a possible interaction between pH value and temperature although the available data should be well designed for this purpose, which are not the case for the data in this study. So, the neural network model has a big advantage over other models, which usually account for a single predicted variable.



Fig. 4 Percentage of correctly predicted optimum pH value and optimum temperature by different amino-acid features

(a) Jack Knifing of delete-1 glucosidase, (b) 13-fold cross-validation, (c) 3-fold cross-validation.

• \bullet : pH by $\sigma_1 \times \text{No.}$, $-\bullet$: pH by $\sigma_\alpha \times \text{No.}$, $-\bullet$: pH by f (i)× No., $-\bullet$: pH by DP, $-\circ$: Tm by $\sigma_1 \times \text{No.}$, $-\overline{\sim}$: Tm by σ_α × No., $-\overline{\Box}$: Tm by f(i)× No., $-\diamond$: Tm by DP.

pH: optimum pH value, Tm: optimum temperature, No.: amino-acid composition, σ_1 : inductive effect scale, σ_{α} : normalized polarizability index, f(i): frequency of the 1st residue in turn, DP: amino-acid distribution probability, \times : multiplication.

Actually, the prediction of optimal working condition for enzymes is an understudied area, thus it is important to develop methods along this line of studies. Experimentally and practically, it is impor-广西科学 2011 年 8 月 第 18 卷第 3 期 tant to develop methods to use as simple information as possible to predict the optimal working condition for enzymes.

For an experimentalist, it would be easier to measure optimum pH value as well as optimum temperature than to predict. However, it is only the model that can provide the basis for generalization. Moreover, the model would provide the basis for simulation of catalytic reaction using computer. Thus, our study can be considered as a small step towards such direction.

The results suggest that the amino-acid distribution probability appears better than other features of amino acids, which is reasonable because it is mainly related to amino-acid spatial distribution. Nevertheless, more studies are needed in order to better predict the optimal working conditions in different enzymes.

A cknowledgements

The authors wish to thank the Library of Guangxi Zhuang Autonomous Region for purchasing the book, Biometry.

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科学家研制新型催化剂让二氧化碳变成低成本液态燃料

太阳是地球上主要的能量来源,更好地利用丰富的阳光是所有新能源专家试图摘取的"圣杯"。科学家很 早就知道如何将水和二氧化碳转变为氢气和一氧化碳。但是如何高效、批量、而且低廉地转换一直困扰着科学 家。其中的一个拦路虎",是转换过程需要昂贵且稀有的铂或铱等元素来作催化剂,以促使反应发生。最近科 学家将目光投向了二氧化铈,金属铈的氧化物二氧化铈常用于自洁烤箱内壁,可作催化剂使用,铈储量丰富,转 换成本低。

经过反复实验尝试,科学家们研究开发出一种太阳能反应器。该太阳能反应器采用低成本的二氧化铈作为催化剂集中太阳的热量,当将二氧化铈加热至约 1500 摄氏度高温时,会自动地从其结构内释放出氧气;接着将其冷却,氧气离开后留下的空白需要新氧气来填满。在约为 900 摄氏度的较低温度时,铈、氢气和碳都需要氧气,但是铈的需求更强烈,于是,它就会从水和二氧化碳中"掠夺"氧气来填满这些空白,因此,水和二氧化碳就变成了氢气和一氧化碳。大量的氢气和一氧化碳结合在一起可形成液态燃料,为汽车、手提电脑和全球定位系统(GPS)供电。

但是,目前这个将太阳光、二氧化碳和水转变为液态燃料的反应器的转换效率不足1%。科学家表示,理 论上反应器的转换效率可达15%以上。此外,科学家也希望能找到比二氧化铈更好的燃料,降低发生反应所 需要的高温和低温。

(据科学网)